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**Tumor microenvironment-manipulated radiocatalytic sensitizer
based on bismuth heteropolytungstate for radiotherapy enhancement**

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Radioresistance resulted from the intrinsic features of tumors often gives rise to unsatisfied therapeutic outcome. In particular, the tumor microenvironment (TME) with abundant antioxidants, elevated hydrogen peroxide (H_2O_2) and hypoxia has been believed as a tremendous obstacle for radiotherapy. Therefore, developing an effective radiosensitizer in response to both X-ray and the TME is highly imperative but remains a challenge so far. Here, we for the first time explore bismuth heteropolytungstate (BiP_5W_{30}) nanoclusters as radiosensitizers for the TME-manipulated enhancement of radiotherapy. On the one hand, BiP_5W_{30} nanoclusters can increase radiation dose deposition within tumors by high-Z elements like Bi and W. On the other hand, in virtue of the unique electron structure and multi-electron property, they have the capability of depleting glutathione (GSH) via redox reaction and catalyzing the decomposition of H_2O_2 to HO^\bullet to enhance ROS generation upon X-ray radiation. Moreover, reduced graphene oxide (rGO) coupled with BiP_5W_{30} can further improve radiocatalytic activity through promoting electron-hole separation. Simultaneously, due to the considerable near-infrared absorption of rGO, photothermal therapy can overcome the tumor hypoxia microenvironment and thus synergize with radiotherapy. In addition to providing a promising radiosensitizer, this finding is expected to extend the application of polyoxometalates used in the biomedical field.

Keywords: polyoxometalates, radiocatalytic sensitizer, glutathione depletion, reactive oxygen species generation, tumor microenvironment-manipulation

1. Introduction

Radiation therapy (RT), as a major modality of cancer therapy, is a local treatment and can effectively suppress tumor non-invasively.[1, 2] To date, tremendous effort has been devoted to improve the radiotherapeutic efficiency via the internal (a radioactive material) or external (high-energy X-rays) strategies.[2-6] However, radioresistance resulted from the intrinsic features of tumors often gives rise to unsatisfied therapeutic outcome and subsequent tumor relapse. In particular, the specific microenvironment of tumor has been believed as the major obstacle of cancer therapy.[7-9] A common concern is the oxygen-deficient microenvironment in solid tumors largely due to the distorted blood vessels and the fast growth of cancer cells,[10] which could develop systemic acquired resistance to radiation and make cancer cells 2-3 times more radioresistant than normoxic tumors.[11,12] Moreover, because the elevation of antioxidants is a general feature of cancer phenotype, antioxidant system (AOS) within cancer cells may prevent them against the damage from radiation via quenching excessive free radicals.[13] For example, the level of glutathione (GSH) within cancer cells is ~4-fold higher than that of the normal cells[14, 15], thus causing the scavenging of reactive oxygen species (ROS) before their arrival at target sites and subsequent great reduction of radiotherapeutic efficiency.[16, 17] Therefore, it is highly desirable to reduce the intracellular GSH level to circumvent tumor resistance and improve RT efficacy.[17] In addition, hydrogen peroxide (H_2O_2) within a tumor is also over-produced due to the insufficient blood supply and rapid metabolism of cancer cells.[18-20] Notably, it has been

approved that ionizing radiation can increase the generation of superoxide ion ($O_2^{\bullet-}$) by promoting mitochondrial electron transport chain (ETC) and then these free radical are quickly converted to H_2O_2 with the aid of superoxide dismutase (SOD).[21, 22] This further elevates H_2O_2 level in cancer cells during the radiotherapy treatment.[23] Since the above mentioned characteristics of the tumor microenvironment (TME) are rather unique compared to normal tissues, it is conceivable that, if the TME can be manipulated by the internal and external stimuli, which points at the improvement of hypoxia, disturbance of AOS, and conversion of overproduced H_2O_2 into highly toxic hydroxyl radical (HO^{\bullet}), the tumor-specific radiotherapy will be achieved.

Here, for the first time, we show that bismuth heteropolytungstate $[BiP_5W_{30}O_{110}]^{12-}$ (BiP_5W_{30}) nanoclusters, which possess ultrasmall size, well water solubility and versatile properties can serve as a simple yet powerful radiosensitizer in response to both external X-ray and the internal TME. **Figure 1** demonstrates the main therapeutic principles. First, due to the presence of multiple high Z elements such as Bi and W, BiP_5W_{30} nanoclusters could physically increase X-ray radiation dose deposition within cancer cells.[24, 25] Moreover, by virtue of the highest oxidation state of W^{6+} , these nanoclusters have the ability to deplete intracellular GSH pool via redox reaction between W^{6+} and GSH, leading to overwhelm AOS and increase ROS level.[26, 27] Most importantly, as one of the promising photocatalysts, BiP_5W_{30} nanoclusters also have high radiocatalytic activity to produce electron-hole pairs under X-ray irradiation. Subsequently, the radio-generated electrons could be transferred to abundant H_2O_2 in the TME, finally yielding HO^{\bullet} , one of the most

harmful free radicals.[28] Furthermore, to improve radiocatalytic activity for ROS generation, we couple BiP₅W₃₀ nanoclusters with reduced graphene oxide (rGO) to form a heterostructure which is beneficial to promote the electron-hole separation, consequently enhancing the generation of ROS.[29] Besides, the mild photothermal effect caused by rGO under near-infrared (NIR) irradiation is capable of enhancing the intratumoral blood flow, subsequently increasing the concentration of oxygen around the TME. This further decrease the hypoxia-associated radiotherapy resistance and result in the enhancement of radiotherapeutic efficacy.[30] Therefore, the BiP₅W₃₀ nanocluster-based radiocatalytic sensitizer with dual response to both X-ray and the TME could not only enhance local X-ray dose deposition within cancer cells but also manipulate the TME for reversing tumor radio-resistance through the depletion of GSH level, radiocatalytic conversion of H₂O₂ into HO[•], as well as improvement of oxygen level within tumors.

2. Experimental section

2.1 Materials

All of the reagents were analytical grade and used without any further purification. Sodium tungstate dihydrate (Na₂WO₄•2H₂O, 99.00%), bismuth nitrate pentahydrate (Bi(NO₃)₃•5H₂O, 99.99%), potassium acetate (K(CH₃COO), 99.00%) and methanol (CH₃OH) were supplied by Alfa Aesar Reagent Co. Poly(vinyl pyrrolidone) (PVP, M_w≈10 000) was provided by Sigma Aldrich. Phosphoric acid (H₃PO₄) was purchased from Sinopharm Chemical Reagent Co., Ltd. Potassium chloride (KCl, 99.50%) was

obtained from Aladdin Co. Other reagents were all purchased from Beijing Chemical Reagent Co.

2.2 Preparation of BiP₅W₃₀ nanoclusters

K_{12.5}Na_{1.5}[NaP₅W₃₀O₁₁₀]•15H₂O was prepared according to the reported method with an improved yield.[31] Typically, Na₂WO₄•2H₂O (16.5000 g) was dissolved in 15.00 mL of deionized water, followed by the addition of 13.25 mL of 85% H₃PO₄. The resulting mixture was then heated at 120 °C for 12 h under continuous stirring. After slowly cooled down to room temperature, 7.50 mL of deionized water and 5.0000 g of solid KCl were sequentially added to the clear yellow solution, obtaining slightly yellow milk-like mixture. The crude product was filtered, washed with 2 M potassium acetate and methanol, and dried at 60 °C in the vacuum oven for 24 h. To further remove the remaining metal species, the obtained product was dissolved in an appropriate hot-water, making a saturated solution. The solution was then cooled down to room temperature to obtain white pin-shaped crystals. A second recrystallization gave the pure product with the same protocol as described above.

Bismuth heteropolytungstate nanoclusters were synthesized through the ion exchange reaction conducted under hydrothermal conditions. In detail, 2.0000 g of K_{12.5}Na_{1.5}[NaP₅W₃₀O₁₁₀]•15H₂O was dissolved in a 24-mL deionized water containing 0.16-mL HCl, and the solution was heated to 60 °C under vigorous stirring to obtain a homogeneous suspension. Then, 6-mL dispersion of Bi(NO₃)₃•5H₂O (0.2328 g) was dropwise added into the above solution under vigorous stirring. The

obtained mixture was transferred into a 45-mL stainless Teflon-lined autoclave and heated at 160 °C for 12 h. When the autoclave was allowed to cool to room temperature naturally, 8.0000 g of solid KCl was added into the mixture to predicate the product. Finally, the colorless product was collected by filtration, washed several times with ice water, and dried at 60 °C for 24 h.

2.3 Preparation of graphene oxide nanosheets

Graphite oxide was synthesized by a modified Hummer's method.[32, 33] In detail, graphite flakes were oxidized with KMnO_4 in the presence of concentrated H_2SO_4 . After vigorous stirring at 37 °C for 2 h, the appropriate deionized water and 30% H_2O_2 were sequentially added into the mixture. Graphite oxide was achieved by centrifugation and washed several times with deionized water (containing 5% hydrochloric acid). The obtained precipitate was then dispersed in deionized water again and then subjected to dialysis for two weeks to remove any residual metal ion and acid. In order to exfoliate graphite oxide, the product was diluted with deionized water until the final concentration reached to ~1 mg/mL, followed by sonication for 20 h with a probe sonicator. Finally, the obtained dispersion was centrifuged at 12 000 rpm for 10 min, and the supernatant containing graphene oxide (GO) was obtained.

2.4 Preparation of PVP-functionalized hybrid composed of $\text{BiP}_5\text{W}_{30}$ nanoclusters and reduced graphene oxide

The hybrid composed of $\text{BiP}_5\text{W}_{30}$ nanoclusters and reduced graphene oxide (rGO) was prepared via the electrochemical reduction process (ERP).[34] In a typical

synthesis, BiP₅W₃₀ nanocluster (0.2000 g) was dissolved into 0.5 M H₂SO₄ solution (15.32 mL, as the electrolyte solution), followed by the addition of the aqueous dispersion of GO (4.68 mL, ~2.0 mg/mL) under stirring, forming a homogeneous suspension. Then, controlled potential coulometry measurement was performed in a three-electrode conventional glass cell with the mixture of BiP₅W₃₀ nanoclusters and GO. In this system, a glassy carbon (GC) plate was applied as the working electrode, a platinum sheet with large surface area as the counter electrode, and a saturated calomel electrode (SCE) as the reference electrode. More importantly, SCE and platinum sheet must be separated from the electrolyte by a glass frit. After saturation of the electrolyte with ultrahigh-purity argon (Ar) gas for 0.5 h, a potential of -0.62 V vs. SCE was set on the working electrode to fully reduce BiP₅W₃₀ nanocluster by 10 electrons. When the electrochemical reduction was completed, the mixture changed to dark-blue and then stood for several hours until the dark-blue color was disappeared. Then, the suspension was washed several times with ultrapure water and then dialyzed with a dialysis bag (MWCO: 3.5 kDa) for 48 h to remove free ions. (named as PG) To improve the biocompatibility and dispersibility of the synthesized PG, 0.0025 mg of PVP was added into 10 mL of PG (in term of GO, 1 mg/mL). After vigorous stirring for 24 h, the product was collected by configuration, washed several times with deionized water, and finally stored at 4 °C. (named as PVP-PG)

2.5 Photocurrent Measurement

The interdigitated electrodes (IDE) of gold/chrome (Au/Cr) device were used to measure the X-ray-induced photocurrent. Typically, the as-prepared BiP₅W₃₀ or

PVP-PG deposited on the IDE with 20- μm electrode gap. (0.2 mg/mm^2) Then, photoresponse on Au/Cr IDE was performed by using an X-ray tube with the voltage of 50 kV and the current of 75 μA for all experiments. (Mini-X, AMPTEK Inc.) Both X-ray-induced and dark currents from the $\text{BiP}_5\text{W}_{30}$ or PVP-PG were measured by using the following conditions: 80 s period of on/off X-ray by six times, 10-V bias voltage.

2.6 Electrochemical measurement

Electrochemical measurement was conducted in the acidic electrolyte (0.5 M H_2SO_4) and controlled using a CHI 660D electrochemical workstation (CHI Instrument, Inc.). A standard three-electrode setup was used in all experiments: reference electrode (SCE), working electrode (an indium-tin oxide, ITO), and a counter electrode (a platinum sheet). When 0.1000 g of $\text{BiP}_5\text{W}_{30}$ was dispersed into 10-mL 0.5 M H_2SO_4 , the electrolyte was thoroughly degassed with Ar gas for 0.5 h and then flushed with Ar gas to prevent oxygen from dissolving into the electrolyte. The electrochemical measurement was carried out with a scan rate of 10 mV/s.

2.7 Detection of hydroxyl radical

The radio-catalytic generation of hydroxyl radical (HO^\bullet) by PVP-PG under X-ray irradiation was examined by the HO^\bullet -specific indicator[35] (terephthalic acid, TA, 5 mM, Alfa Aesar), which can react with HO^\bullet to form 2-hydroxy terephthalic acid (TAOH, the maximum fluorescence peak at 435 nm). The solution of TA was first added into the control, PVP-PG, X-ray, and PVP-PG+X-ray, respectively, in the absence or presence of H_2O_2 (1 mM). After irradiation with/without X-ray (50 kV, 75

μA , 10 min), the mixtures were gently shaken at 37 °C for 12 h in the dark, and then changes of fluorescence emission at 435 nm were recorded. The method of active species trapping experiment is similar to the radio-activity test, but the electron scavenger ($\text{Na}_2\text{S}_2\text{O}_8$, 0.2 mM) is added.

2.8 Detection of hydrogen peroxide *in vitro*

For the detection of relative content of H_2O_2 within HeLa cells, cells were seeded in 6-well plate at a density of 5×10^5 cells/well with complete medium for 24 h, and then treated with or without PVP-PG (12 $\mu\text{g}/\text{mL}$). After incubation for 6 h, different groups were irradiated with or without X-ray and then incubated for another 6 h. The cells were collected by centrifugation and washed twice with PBS. With the addition of 100 μL of Lysis Buffer of H_2O_2 , the cells were homogenized and centrifuged at 12000 g for 4 min at 4 °C. Finally, 50- μL supernatants and 100- μL test solutions (hydrogen peroxide assay kit, Beyotime) were mixed in 96-well plate, placed at room temperature for 0.5 h, and measured immediately with a microplate spectrophotometer at the wavelength of 570 nm.

For the confocal fluorescence imaging, HeLa cells were seeded in glass bottom confocal dishes at a density of 2×10^5 cells/dish with complete medium for 24 h, and then incubated in the presence or absence of PVP-PG (12 $\mu\text{g}/\text{mL}$). After 6 h, different groups were treated with or without X-ray and incubated for another 3 h. Then, cells in different groups incubated with mixture solution for 30 min according to the protocol of assay kit (Fluorimetric Hydrogen Peroxide Assay Kit Infra-red (IR)

Fluorescence, Sigma-Aldrich). When treated with Hoechst for 15 min, the cells were monitored by confocal microscope.

2.9 Detection of hydroxyl radical *in vitro*

To determine the formation of hydroxyl radical, specific hydroxyl radicals ELISA assay (Sensitivity: 0.10 ng/mL) was used to detect the intracellular generation of HO[•]. [36] The kit applies double-antibody one-step sandwich enzyme-linked immunosorbent assay (ELISA). The test antibody and the standard are incubated together with HRP-labeled detection antibody in human hydroxyl radical (HO[•]) capture antibody pre-coated microplates. The microplates are incubated and then washed thoroughly. Then, the microplates are incubated with a substrate TMB for HRP enzyme, which forms a blue colored complex by the enzyme-substrate reaction. Finally, a stop solution is added to stop the above reaction, turning the solution yellow. A microplate reader is used to spectrophotometrically measure the intensity of color (450 nm). Because HO[•] from HO[•]-HRP conjugate and samples compete for the anti-HO[•] antibody binding site, the intensity of the color is inversely proportional to the HO[•] concentration. HeLa cells were plated in 6 well plates at 3×10^5 cells/well overnight and then the cells were pretreated with PVP-PG (12 µg/mL) for 6 h. After that, the cells were irradiated with X-ray for 15 min and then incubated for another 2 h. Then the cells were washed three times with PBS and collected in 1.5 mL centrifuge tube. In an attempt to dissociate the cells and release endogenous components, the cells were lysed with a RIPA lysis (Beytime) and centrifugated at 1000 g for 20 min at 2-8 °C. Finally, the supernant of different groups were collected

for Human HO[•] ELISA Kit assay according to the manufacturer's instructions (Shang Hai Jianglai Biological Technology Co., Ltd.).

2.10 Ellman's Assay

Glutathione (GSH) oxidation was examined by Ellman's assay and all experiments were conducted in the dark. Ellman reagent (5, 5'- dithiobis (2-nitrobenzoic acid), DTNB, 100 mM, Alfa Aesar) is able to cleave the disulfide bonds (-S-S-) to obtain a yellow product (2-nitro-5 thiobenzoate acid). Typically, the dispersion of PVP-PG (BBS, 100 µg/mL, 225 µL) was mixed with the GSH solution (1.0 mM, 225 µL). The control group is GSH solution. After irradiation with/without X-ray for 10 min, the mixtures were incubated at 37 °C for 0.5, 1, 1.5, 2, 4, and 6 h, respectively. Afterward, 785 µL of 0.05 M Tris-HCl (pH=8.8) solution and 15 µL bicarbonate buffer solution (50 mM, pH=8.77) with DTNB (100 mM) were added into the above mixtures to react for 5 min. After centrifugation at 12 000 rpm for 10 min, the absorbance of supernatants at 410 nm was recorded on a microplate spectrophotometer (Multiskan MK3, Thermo fisher Scientific, England).

2.11 Detection of reactive oxygen species

The catalytic activity of PVP-PG under X-ray irradiation was assessed using a reactive oxygen species (ROS) probe 2',7'-dichlorodihydrofluorescein diacetate (H₂DCF-DA, 10 µM, Beyotime Co, Sigma-Aldrich, USA)[37] which can be deacetylated to form H₂DCFH and oxidized to form a highly fluorescent 2',7'-dichlorofluorescein (DCF) by ROS. Firstly, 0.5-mL stock solution of H₂DCF-DA in dimethyl sulfoxide (DMSO) was mixed with 2-mL 0.01 M NaOH in

the dark. After stirring for 0.5 h at room temperature, the reaction was stopped by adding 10-mL phosphate buffer saline (PBS, 25 mM, pH7.20). Finally, the resulting solution contained 10 μ M H₂DCFH, and was wrapped by aluminum foil and kept on ice before use. Then, PBS, PVP-PG, X-ray, and PVP-PG+X-ray were mixed with H₂DCFH solution, respectively. After irradiation with X-ray for 15 min, the dispersion of individual group was centrifuged at 12 000 rpm and the fluorescence of the supernatant was measured by a fluorescence spectrophotometer (Horiba FluoroLog-3, Japan).

2.12 Detection of ROS generation *in vitro*

HeLa cells were incubated in glass bottom confocal dishes at a density of 2×10^5 cells/dish with complete medium for 24 h, and then treated with or without complete medium containing BiP₅W₃₀ nanoclusters (80% loading on rGO, 10 μ g/mL) and PVP-PG (12 μ g/mL). The treated cells were washed twice with PBS and then incubated with H₂DCFH-DA containing RPMI medium for 20 min. When washed with fresh medium, groups were treated with or without near-infrared (NIR) light (808 nm laser, 1 W/cm²) for 10 min and/or X-ray (50 kV, 75 μ A) for 15 min. Finally, the cells were treated with Hoechst 33342 (Beyotime) for 15 min and the fluorescence images of the cells were monitored by the confocal microscope (A1/LSM-Kit, Nikon/PicoQuant GmbH, Japan/Germary).

2.13 Study of effect of GSH level on radiotherapeutic outcome

HeLa cells were incubated in a 96-well plate at a density of 5×10^3 cells per well for 24 h. To regulate the level of GSH, the cells were first treated with 500 μ M lipoic acid

(LPA, a GSH synthesis enhancer) for 24 h and 2 μ M N-methylmaleimide (NMM, a GSH scavenger) for 20 min, respectively. Then, cells were incubated with PVP-PG (12 μ g/mL) for 6 h, and subsequently irradiated with X-ray (50 kV, 75 μ A) for 15 min. After 24-h incubation, cell viability of HeLa cells was assayed by CCK-8 assay by microplate spectrophotometer.

2.14 Determination of GSH/GSSG ratio *in vitro*

The total glutathione and oxidative glutathione were determined by GSH/GSSG assay kit (Beyotime).[37] For intracellular determination, HeLa cells were incubated in 6-well plate at a density of 1×10^5 cells/well for 24 h and then incubated with PVP-PG (12 μ g/ml) for 12 h. The treated cells were irradiated with X-ray for 15 min and then incubated for another 6 h. The cells were collected by centrifugation at 12 000 rpm for 5 min at 4 $^{\circ}$ C and added with $3 \times$ protein solution M. After vortex sufficiently, the mixed solutions were frozen and thawed twice using liquid nitrogen and 37 $^{\circ}$ C water, respectively. The samples were placed at 4 $^{\circ}$ C for 5 min and then centrifuged at 10 000 rpm for 10 min at 4 $^{\circ}$ C. Finally, the supernatant was used for GSH and GSSG assay, and GSH/GSSG was determined according to the manufacturer's instructions.

2.15 *In vitro* colony formation assay

Colony formation assay is an assay of the survival ability of a single cell to grow into a colony *in vitro*. HeLa cells with different numbers (125, 250, 500, 1000, and 2000 cells) were incubated in a 6-well plate with complete medium for 24 h. After attachment, the cells were treated with complete medium containing PVP-PG (12

$\mu\text{g/mL}$). After incubation for 6 h, the cells were treated with 808-nm laser (1 W/cm^2) and/or X-ray (0, 2, 4, 6, and 8 Gy) and incubated with complete medium for 10 d. Finally, colonies were stained with Giemsa dye (Beijing Solarbio Science & Technology Co., Ltd) and evaluated the effects of different treatments with the survival fraction.

2.16 *In vitro* DNA double-strand breaks assay

HeLa cells were seeded on cover glass and then incubated in 24-well plate at a density of 3×10^4 cells/well. After attachment for 24 h, the cells were treated with or without complete medium containing PVP-PG ($12 \mu\text{g/mL}$) and then treated with 808-nm laser (1 W/cm^2) and X-ray (50 kV, $75 \mu\text{A}$, 10 min). The treated cells were fixed with 4% paraformaldehyde ($500 \mu\text{L/cell}$) for 10 min after 30 min. When washed three times with PBS, the cells were blocked in 5% FBS and 1% tritonX-100 in PBS ($300 \mu\text{L}$) for 1 h, and then stained with anti- γ -H2AX antibody (Proteintech Group Inc. Co., Ltd., China) containing 5% FBS and 1% tritonX-100 overnight at $4 \text{ }^\circ\text{C}$. Finally, the cells were incubated with the secondary antibody anti-rabbit AlexaFluor-488 conjugated IgG (Abcam) for 1 h and then stained with Hoechst for 15 min. Images were visualized with the confocal microscope. A semiquantitative analysis was used to evaluate the number of foci per cell expressing the γ -H2AX markers.

2.17 *In vitro* apoptosis analysis by flow cytometry

The apoptosis of the cells was detected using the Annexin V-FITC apoptosis detection Kit (Beyotime). Briefly, HeLa cells were seed in 24-well plates with a density of 5×10^4 cells/well for 24 h. After treated with or without PVP-PG (12

$\mu\text{g/mL}$), different groups were irradiation with or without an 808-nm NIR laser (808 nm laser, 1 W/cm^2) and X-ray (50 kV, $75 \mu\text{A}$, 10 min). Then, cells were collected by trypsinization (without EDTA), washed twice with PBS, re-suspended in Annexin V binding buffer and further incubated in Annexin V and PI at room temperature for 15 min in the dark. Finally, cells were subjected to flow cytometry analysis (BD Accuri C6, USA).

2.18 Photothermal therapy/radiotherapy synergistic treatment *in vivo*

For *in vivo* therapy, seven randomly divided groups of HeLa tumor-bearing mice were employed for photothermal therapy/radiotherapy synergistic performance: (I) Control, PBS; (II) NIR; (III) PVP-PG; (IV) PVP-PG+NIR; (V) X-ray; (VI) PVP-PG+X-ray and (VII) PVP-PG+NIR+X-ray. Once the tumors attained 80-100 mm^3 , mice were administrated with 20- μL solution of PBS (groups I, II and V) or PVP-PG (2 mg/mL , other groups) via intratumor injection, respectively. Group (II) and (IV) was treated with 808-nm NIR laser (0.35 W/cm^2 , 10 min), groups (V) and (VI) were treated with X-ray (6 Gy), and group (VII) was first treated with 808-nm NIR laser (0.35 W/cm^2 , 10 min) and then X-ray (6 Gy). The temperature profiles were recorded in real time using a thermal camera during laser irradiation treatment. Following each treatment, the data of body weight and tumor volumes in each group were recorded over the course of 25 d.

2.19 Blood and histology examinations

Each group with five mice at different treatments were euthanatized on 3rd and 25th day after administration. Standard procedures for blood routine examination were

performed with blood obtained from mice on 25th day. In addition, the tumor, heart, liver, spleen, lung, and kidneys were collected for hematoxylin and eosin staining (H&E). Images were collected using an inverted fluorescence microscope.

2.20 *In vivo* DNA damage study

DNA damage was evaluated semi-quantitatively by γ -H2AX. First, tumors were harvested and fixed in 10% formalin followed by paraffin embedding. Then, the staining was carried out on tissue sections. To quantify the toxicity by γ -H2AX, the primary antibody and secondary antibody were utilized. Images were analyzed using an inverted fluorescence microscope.

3 Results and discussion

3.1 Synthesis and characterization of PVP-PG

Figure 2a shows how we combine the significant features of rGO and BiP₅W₃₀ nanoclusters to form a simple yet efficient radiosensitizer. First, stable BiP₅W₃₀ nanoclusters were successfully synthesized according to the previous method,[31] as demonstrated by the results of Fourier transform infrared (FT-IR) and X-ray photoelectron spectroscopy (XPS). (**Figure S1-2**) In these nanoclusters, five PW₆O₂₂ units derived from the Keggin anion [PW₁₂O₄₀]³⁻ arrange in a crown and bismuth ion is positioned in it.[31] (**Figure S3**). From ultraviolet-visible analysis, the structure of BiP₅W₃₀ nanoclusters was rather stable in various solutions. (**Figure S4-5**) Then, these nanoclusters were anchored on the surface of graphene nanosheets through an electrochemical method,[34] obtaining BiP₅W₃₀/rGO hybrid (PG). (**Figure S6a-h**) In

particular, the content of $\text{BiP}_5\text{W}_{30}$ nanoclusters coupled with rGO could be fine-controlled by changing the initial ratio of rGO and $\text{BiP}_5\text{W}_{30}$ nanoclusters in the dispersion. **(Figure S6i)** During this process, graphene oxide (GO) was reduced to form high conductive rGO that is capable of enhancing radiocatalytic activity.[38] **(Figure S7-8)** Remarkably, the thickness of rGO nanosheets (~ 0.5 nm) increases to ~ 1.5 nm, which indicates the successful loading of $\text{BiP}_5\text{W}_{30}$ nanoclusters on the rGO surface. **(Figure 2b, Figure S9)** Finally, to improve the colloidal stability and biocompatibility, PG was further functionalized with PVP,[39] named as PVP-PG. **(Figure 2g, Figure S10)** Transmission electron microscope (TEM) image shows that $\text{BiP}_5\text{W}_{30}$ nanoclusters are still evenly distributed onto the surface of graphene nanosheets, in accordance with Raman spectrum analysis. **(Figure 2c, Figure S11)** Energy dispersive X-ray spectroscopy (EDS) analysis demonstrates the homogeneous distribution of Bi, P and W elements on the surface of rGO, which is also verified by XPS analysis. **(Figure 2d-f, Figure S12)** Moreover, PVP-PG possesses a high zeta potential value (-35 mV) and exhibits exceptional stability in biofluids and good biocompatibility. **(Figure 2h, Figure S13)**

3.2 Radiocatalysis effect of PVP-PG

Polyoxometalates (POMs), the well-defined metal-oxygen clusters, have been approved to possess desirable reversibility in multi-electron redox reaction, which enables them with attractive photocatalytic properties.[40, 41] Similar to semiconductor nanoparticles, the electron structure of POMs consists of the highest occupied molecular orbital mainly localized on oxygen atoms (HOMO, like valence

band in semiconductor) and the lowest unoccupied molecular orbital mainly localized on d^0 or d^1 transition metal atoms (LUMO, like conductive band in semiconductor).[42, 43] When POMs are irradiated by an appreciate light (usually in ultraviolet region) with energy equal to or higher than their gap (HOMO-LUMO gap), electrons could be excited from HOMO to LUMO, generating highly reactive electron-hole pairs.[41] Subsequently, these excited electrons and holes are trapped in surface sites and react with surrounding water and/or other molecules via either reductive (electron) or oxidative (hole) pathways.[44] Consequently, the generated free radicals can cause indirect damages to cellular components like DNA, proteins and lipids.[45] It is worth noting that if the energy level of POMs matches with the potential of H_2O_2 , they could catalyze low-toxic H_2O_2 to generate HO^\bullet . [28] This offers an opportunity to use overproduced H_2O_2 within the TME to improve the therapeutic effects. However, due to the strong light scattering and absorption by skin and tissue, the low-energy photons (from ultraviolet to visible regions) cannot penetrate into human body. This greatly limits their application to eradicate deep located tumors.[46, 47]

Recently, several attempts have been devoted to fabricate high Z semiconductors capable of generating ROS under X-ray irradiation.[48, 49] Inspired by this trend, we for the first time explore the possibility of using PVP-PG as an X-ray triggered photocatalyst for RT enhancement because of its well radiocatalytic property and high X-ray absorption ability. The optical band gap of PVP-PG was evaluated by ultraviolet-visible diffuse reflectance spectroscopy (DRS). (**Figure 3a**) PVP-PG exhibits strong absorption in the ultraviolet region. Based on the Tauc plot of

transformed Kubelka-Munk function versus the photon energy, the optical band gap (E_g) of PVP-PG was thus determined to be 2.76 V.[50, 51] Then, the photocurrents generated from both $\text{BiP}_5\text{W}_{30}$ and PVP-PG shows rapid response during X-ray switching on/off, indicating the electron-hole separation.[49, 52] **(Figure 3b)** Notably, the $\text{BiP}_5\text{W}_{30}/\text{rGO}$ heterojunction structure (PVP-PG) is able to generate significant photocurrent that is about 4-fold higher than that of single $\text{BiP}_5\text{W}_{30}$ nanoclusters. The higher photocurrent response of PVP-PG to X-ray indicates more efficient separation of radio-generation electron-hole pairs due to rGO with two-dimensional planar structure which can facilitate electron transportation.[29] As a consequence, these radiogenerated electrons and holes are able to initiate the chemical reactions because of the strong photo-oxidative ability of the holes and photo-reductive ability of the electrons.[41] Encouraged by this, we used the energy levels of PVP-PG to predict whether the abundant components like H_2O_2 or H_2O in the TME can react with electrons and/or holes. Based on the fact that redox potential of first reduction peak corresponds to the energy of the LUMO level (mainly the W atoms), the value of LUMO is measured to be 0.10 V vs SHE (standard hydrogen electrode) by cyclic voltammetry.[53] **(Figure 3c)** The energy level of HOMO is thus estimated at 2.86 V from $E_g + E_{\text{LUMO}}^\theta$. Considering the acidic pH of the TME, all these values were converted to electrochemical potentials at pH 5.00 and 6.50 using Nernst equation.[54] **(Table S1)** According to the results achieved above, we further investigated electron transfer process for radiocatalytic reactions. **(Figure 3d)** Under X-ray irradiation, the absorption of energy by $\text{BiP}_5\text{W}_{30}$ can excite electrons from HOMO to LUMO to rGO,

leaving positive holes behind. As calculated, the LUMO energy level of PVP-PG is lower than the potential of $O_2/O_2^{\cdot-}$, but higher than that of H_2O_2/HO^{\cdot} . Hence, the radio-generated electrons could not be accepted by O_2 , but caught by intracellular H_2O_2 for an oxidation reaction to form HO^{\cdot} . In parallel, since the HOMO energy level of PVP-PG is lower than the potential of HO^{\cdot}/H_2O , the corresponding holes also diffuse to the surface and accept by H_2O for a reduction reaction to produce HO^{\cdot} . Therefore, the main contribution of the resulting ROS generated by PVP-PG may be HO^{\cdot} .

To examine HO^{\cdot} generation from radiocatalytic reactions, the HO^{\cdot} -specific indicator was used in the presence of PVP-PG and/or H_2O_2 . There is almost no fluorescence without X-ray irradiation in the presence of PVP-PG. With X-ray irradiation, the fluorescence of 2-hydroxy terephthalic acid[35] (TA-OH, produced by the specific reaction of terephthalic acid with HO^{\cdot}) in the presence of PVP-PG increases 1.41-fold higher compared to the control without PVP-PG. (**Figure 3e**) Moreover, the addition of H_2O_2 leads to 1.76-fold fluorescence enhancement under X-ray irradiation. It is clear that both H_2O_2 and H_2O are able to capture radio-generated electrons and holes, respectively, to form high active HO^{\cdot} . This is further demonstrated by the trapping experiment which was conducted in the presence of $S_2O_8^{2-}$ scavenger (It can react with electron to yield SO_4^{2-}). [42] (**Figure 3f**) To further verify radio-catalytic activity of PVP-PG, the *in vitro* HO^{\cdot} generation was evaluated. Given that the intracellular H_2O_2 was the source of HO^{\cdot} , its content was first determined by hydrogen peroxide assay kit. Remarkably, H_2O_2 level significantly

increases upon X-ray irradiation, which is ascribed to ionizing radiation as mentioned before. **(Figure 3g)** It is noted that H_2O_2 level also increase after incubation with PVP-PG, which may be attributed to the depletion of intracellular GSH pool.[55] However, it clearly decreases under X-ray irradiation in the presence of PVP-PG, indicating the feasible consumption of H_2O_2 by radio-catalytic reaction. This can be further demonstrated by fluorimetric hydrogen peroxide assay kit which can specifically measure hydrogen peroxide. **(Figure 3h)** Furthermore, the content of HO^\bullet within cancer cells was directly detected by specific hydroxyl radicals ELISA assay.[36] **(Figure 3i)** Once combining PVP-PG with X-ray irradiation, the concentration of HO^\bullet shows about 1.30-fold higher compared to the control. This demonstrates the speculation of radio-catalytic effect and the foregoing results. Therefore, both the theoretical and experimental results confirm that PVP-PG has the ability to convert non-toxic H_2O and low-toxic H_2O_2 into highly active HO^\bullet under X-ray irradiation, which contributes to the enhancement of RT.

3.3 Redox effect of PVP-PG

Apart from the radiocatalytic ability to enhance ROS generation, PVP-PG also has the potential for depleting GSH. As we mention before, GSH is critical to protect against ROS and plays an important role in radiation and chemical drug resistance of cancer cells.[56] Depletion of overexpressed GSH within cancer cells could increase the amount of ROS arrived at the target sites to enhance the therapeutic outcome.[37, 57, 58] For BiP_5W_{30} , multi-electron redox activity of transition metal W^{6+} makes a possibility of electrons transfer reactions between GSH and W^{6+} , leading to consume

the intracellular GSH. From the XPS results of elemental states of W atoms, two peaks arising from 4f_{5/2} and 4f_{7/2} are observed, evidencing the characteristic +6 oxidation state. **(Figure 2f)** As expected, the potential of GSSG/GSH couple is higher than the energy level of W⁶⁺ (LUMO), which makes electrons thermodynamically favorable transfer from GSH to W⁶⁺ and leads to redox reaction to deplete GSH. **(Figure 4a)** The content of GSH under various treatments was quantitatively measured by using Ellman's assay. **(Figure 4b-c)** The control with only GSH in buffer has no change even after 6 h, while GSH is gradually consumed in the presence of PVP-PG, indicating the occurrence of redox reaction, but not some physical changes. To investigate the influence of GSH on ROS generation, detection of ROS under X-ray irradiation with or without GSH was monitored and quantified by 2',7'-dichlorodihydrofluorescein diacetate (H₂DCFDA) which can measure the total amount of ROS, including H₂O₂, HO[•], and O₂^{•-}. **(Figure 4d)** Note that there is almost no fluorescence without X-ray irradiation. Under X-ray irradiation, it is found that the fluorescence of the indicator is remarkably enhanced in the control. This is not surprising because ionizing radiation is known to produce ROS through water radiolysis. More interestingly, though the cells treated with PVP-PG plus X-ray irradiation show lower H₂O₂ level than those treated with PVP-PG only, far more HO[•] could be produced from H₂O₂ and H₂O under X-ray irradiation, resulting in the highest level of ROS among these groups. However, the fluorescence intensity dramatically decreases in the presence of GSH, indicating that GSH could indeed deplete ROS to a large extent. Importantly, the fluorescence enhancement (F/F₀) of

PVP-PG under X-ray is 1.71-fold higher than that of the control. In the presence of GSH, PVP-PG still could achieve 1.38-fold ROS than that of the control. Moreover, the amount of generated ROS is radiation time-dependent. **(Figure S14)**

To further verify radiocatalytic effect and redox effect of PVP-PG, the *in vitro* ROS generation and GSH depletion was evaluated. Firstly, qualitative image analysis of ROS generation was performed on fluorescence microscopy images of cancer cells pre-treated with H₂DCFDA. As shown in **Figure 4g**, HeLa cells treated with PVP-PG plus X-ray show much higher fluorescence intensity than those only irradiated by X-ray or incubated with PVP-PG. Simultaneously, PVP-PG plus X-ray showed better ability of ROS generation than BiP₅W₃₀ nanocluster plus X-ray, resulting in more excellent radiotherapeutic performance as demonstrated by clonogenic assay. **(Figure S15)** This can be attributed to an increased in the yield efficiency of electron-hole pairs because rGO facilitates electron transportation as mentioned before. Next, to test whether GSH depletion can sensitize cancer cells to ROS, we assessed the cell viability under different treatments in the presence of α -lipoic acid (LPA, a GSH synthesis enhancer) or N-methylmaleimide (NMM, a GSH scavenger).[59] As shown in **Figure 4e**, the viability of HeLa cells treated with PVP-PG and NMM is markedly lower than that of cells incubated with PVP-PG alone, while a distinct increase of cell viability is observed when HeLa cells were treated with LPA. These results clearly indicate that the cytotoxicity of PVP-PG can be potentiated by depleting intracellular GSH. Moreover, the ratio of GSH to oxidized glutathione (GSSG) was determined to verify the redox activity of PVP-PG for overwhelming AOS *in vitro*.[37] **(Figure 4f)**

It is found that PVP-PG cause a significant decrease in the cellular GSH/GSSG ratio compared with the group irradiated by X-ray. Therefore, we can infer PVP-PG is capable of diminishing GSH level, causing the increase of ROS level, and subsequently enhancing the outcome of radiotherapy.

3.4 PTT and RT synergetic therapy *in vitro*

Based on the radiocatalytic activity for ROS generation and redox activity for GSH depletion, we evaluated the radiotherapeutic inhibitory activities of PVP-PG on HeLa cells. First, the result of cell colony formation indicates that PVP-PG can significantly enhance the radiation-induced inhibition of a single cell to grow into a colony.[60] **(Figure 5a)** For instance, the survival fraction of HeLa cells treated by X-ray or PVP-PG alone are 71% and 91%, respectively. **(Figure 5d)** After co-treated with X-ray and PVP-PG, the survival fraction dramatically declined to 25%. The clonogenic survival assay was also performed to evaluate *in vitro* radiotherapy enhancement. **(Figure 5e)** Similarly, the group treated with PVP-PG shows much lower percentage of viable cell colonies than that without PVP-PG treatment, even at the same X-ray irradiation dose. The sensitizer enhancement ratio of PVP-PG plus X-ray irradiation was calculated to be 1.41 using a linear-quadratic model. **(Table S2)** The above results suggest the enhancement of radiation effects of PVP-PG, and similar results were also found in A549 cells **(Figure S16)**. Since ROS-induced DNA damage was the main contribution for radiation enhancement, immunofluorescently labeling for γ -H2AX was conducted to evaluate DNA double-strand breaks.[61] **(Figure 5b)** Consistent with the cell colony assay result, PVP-PG plus X-ray

produces the higher level of DNA damage compared to radiotherapy alone, indicating that X-ray triggered generation of ROS by PVP-PG could remarkably enhance DNA damage and subsequently inhibit proliferation of cancer cells. **(Figure 5f)** In addition, Fluorescein-annexin V and propidiumiodide staining assays suggest that the cellular toxicity is associated with apoptosis and necrosis. It clearly shows that irradiation of PVP-PG with X-ray leads to more dead cells than the control even in 24 h. **(Figure 5c)** Apart from its ability for radiotherapy enhancement, PVP-PG also has a potential for photothermal therapy (PTT) upon NIR laser irradiation because of the considerable absorbance of rGO in the NIR region.[62] **(Figure S17)** From the evaluation of photothermal performance, PVP-PG can be used as an efficient PTT agent with high photothermal conversion efficiency of ~39.10% and excellent photostability. **(Figure S18-20)** Therefore, based on the above findings, we further evaluated the synergistic effect of RT and PTT in the presence of PVP-PG. More encouragingly, the remaining alive cells of clonogenic survival assay dramatically decreased to 5%, suggesting the combined therapy exhibits a remarkable synergistic effect and enhances the therapeutic efficacy. **(Figure 5d)** This result is further confirmed by the analysis of DNA damage and flow cytometric apoptosis assay. Apparently, the combination of RT with PTT could effectively decrease the radiation dose and reduce the side effects of radiation to achieve the same radiotherapeutic effect. **(Figure 5e)**

3.5 PTT and RT synergetic therapy *in vivo*

In attempt to further investigate the feasibility of using PVP-PG as an effective radiosensitizer, *in vivo* therapeutic experiments were carried out on HeLa tumor

xenograft on BALB/c nude mice. PBS (as control) and PVP-PG were administrated intratumorally, respectively, followed by treatments with X-ray and/or 808 nm laser irradiation. Notably, for the group irradiated with 808 nm laser, the temperature of tumors injected with PVP-PG increases to ~ 45 °C, a temperature suitable for mild PTT,[30, 63] while the control is not notably heated (~ 37 °C). (**Figure 6a and Figure S21**) To assess the therapeutic outcome of PVP-PG, the changes of relative tumor volume were recorded every other day for 25 days. (**Figure 6b**) Compared to the control group and the single-treated groups (PVP-PG, X-ray) which showed a rapid progression of tumor volumes, the co-treated (PVP-PG plus X-ray) groups exhibited significant tumor inhibition. This could be further demonstrated by the quantitative antitumor efficacies calculated by the tumor growth inhibition ratio, i.e., PVP-PG (10.21%), X-ray (39.21%) and X-ray plus PVP-PG (81.40%). (**Table S3**) The satisfactory radiosensitizing effect is supposed to be attributed to the manipulation of the TME and enhanced the generation of ROS, leading the death of cancer cells in an apoptosis pathway.[64, 65] Moreover, the combination of RT and PTT produces much enhanced inhibition of cancer tumor growth, indicating the significant *in vivo* synergistic therapeutic effect in radiothermotherapy treatment. This could be attributed to the improvement oxygen level in the TME and thus reduce the hypoxia-associated radioresistance of tumor. As verified by photoacoustic (PA) analysis, compared to the control, the injection of PVP-PG upon NIR irradiation could markedly increase the signal intensity of oxygenated hemoglobin.[66] (**Figure S22**) The similar results shown in the photographs and weights of tumors in all groups

directly exhibited that the growth of HeLa tumor could be effectively inhibited after the administration of PVP-PG followed by X-ray and/or 808 nm laser irradiation. **(Figure 6c-d)** Notably, tumors on mice with injection of PVP-PG were mostly eliminated after both NIR laser and X-ray irradiation, and there was no case of re-growth in our observation period. **(Figure 6e)** Moreover, according to the biodistribution of PVP-PG and BiP₅W₃₀ nanoclusters, it was found the PVP-PG could be gradually cleared out from the body during therapeutic period, thereby sharply reducing the potential side effects. **(Figure S23)** Meanwhile, compared to BiP₅W₃₀ nanoclusters, the combination of rGO and BiP₅W₃₀ nanoclusters can induce higher tumor accumulations and avoid the fast removal from the tumors.

To evaluate the pathological damages to tumors by PVP-PG and further understand the mechanism for the inhibition of tumor growth, the histopathology images of the dissected HeLa tumors are shown in **Figure 6f**. The differential destruction of tumor cells could be observed in the H&E images of HeLa tumor tissues for the group co-treated with PVP-PG plus X-ray and the synergistic group after 3-day post-injection. After 25 days, in contrast to partial damage of X-ray alone, almost all of the cells in these two groups are heavily destroyed during therapeutic period, particularly severe cancer necrosis and late-stage apoptosis featuring karyopyknosis and karyorrhexis. Therefore, immunohistochemical analysis was exploited to evaluate DNA damage in tumor section on 3rd day after treatments. **(Figure 6g)** There is a significant DNA double-strand break on tumor sections in both the group co-treated by PVP-PG plus X-ray and the synergistic group, compared to all other groups. These

further demonstrate the potential ability of PVP-PG to the TME-manipulated radiocatalytic enhancement of radiotherapy and radiothermotherapy. Over the therapeutic period of 25 days, there are no obvious difference in body weights of mice in the control and all therapeutic groups, indicating that non-significant toxicity is induced by PVP-PG. **(Figure S24)** Moreover, H&E staining assays of major organs from all of therapeutic groups, as well as blood routine data show no noticeable tissue damage or adverse effect compared to the control, further certifying the high biocompatibility of PVP-PG during the therapeutic process. **(Figure S25-26)** In addition to the treatment efficacy, PVP-PG with high-Z metals and rGO could also act as a dual-modal imaging contrast agent for CT and PA imaging. **(Figures S27-28)** Thus, combination of radiotherapy with different imaging modalities can make it possible to visualize towards sensitizer delivery, internalization, and comprehensive evaluation of therapeutic outcome.[67]

4. Conclusion

In summary, we demonstrated a bismuth heteropolytungstate-based radiocatalytic sensitizer (PVP-PG) for the TME-manipulated enhancement of radiotherapy, based on the unique electron structure and multi-electron properties that is sensitive to X-ray and the TME. Similar to currently reported radiosensitizers, such as gold, PVP-PG is able to amplify X-ray radiation dose at the tumor sites via Compton and Auger effect.[68] Apart from the above contribution, radioactivable PVP-PG can also be stimulated to bring an extra sensitization via “radiocatalytic effect”, which ingeniously exploits abundant H_2O_2 and H_2O within the TME to generate more toxic

HO[•]. The radiotherapeutic outcome can be further improved in virtue of redox-activity of PVP-PG to deplete GSH for disturbance of AOS via “redox-reaction effect”. Moreover, NIR-triggered PTT based on PVP-PG can overcome tumor hypoxia microenvironment and synergize with radiotherapy. Taken together, all of the above events hopefully facilitate localized energy deposition in the vicinity of PVP-PG as well as the manipulation of the TME, resulting in better therapeutic outcome and fewer sides effects. To the best of our knowledge, this is the first study that employs both catalytic reaction and redox reaction based on POM for radiotherapy enhancement that is induced by the specific features of TME. Overall, this work not only provides new insight into the design of smart radiosensitizers to achieve more precise and effective radiotherapy, but also extends the application of polyoxometalate used in the biomedical field.

Supporting Information

Supplementary data related to this article can be found at ((will be filled in by the editorial staff))

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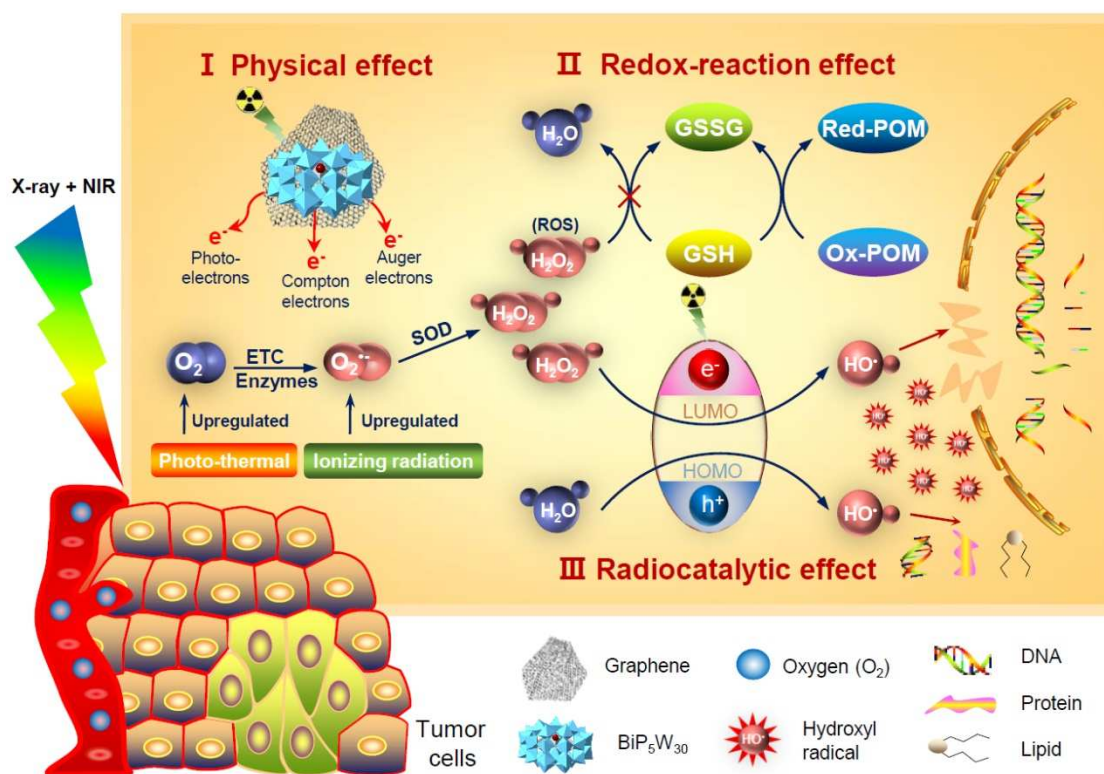


Figure 1. Principle of the TME-manipulated radiocatalytic sensitizer. Schematic representation of mechanism of PVP-PG on the amplification of X-ray radiation dose deposition and the manipulation of TME for reversing tumor radioresistance through depleting GSH concentration, converting H₂O₂ to HO• as well as improving oxygen level in tumors.

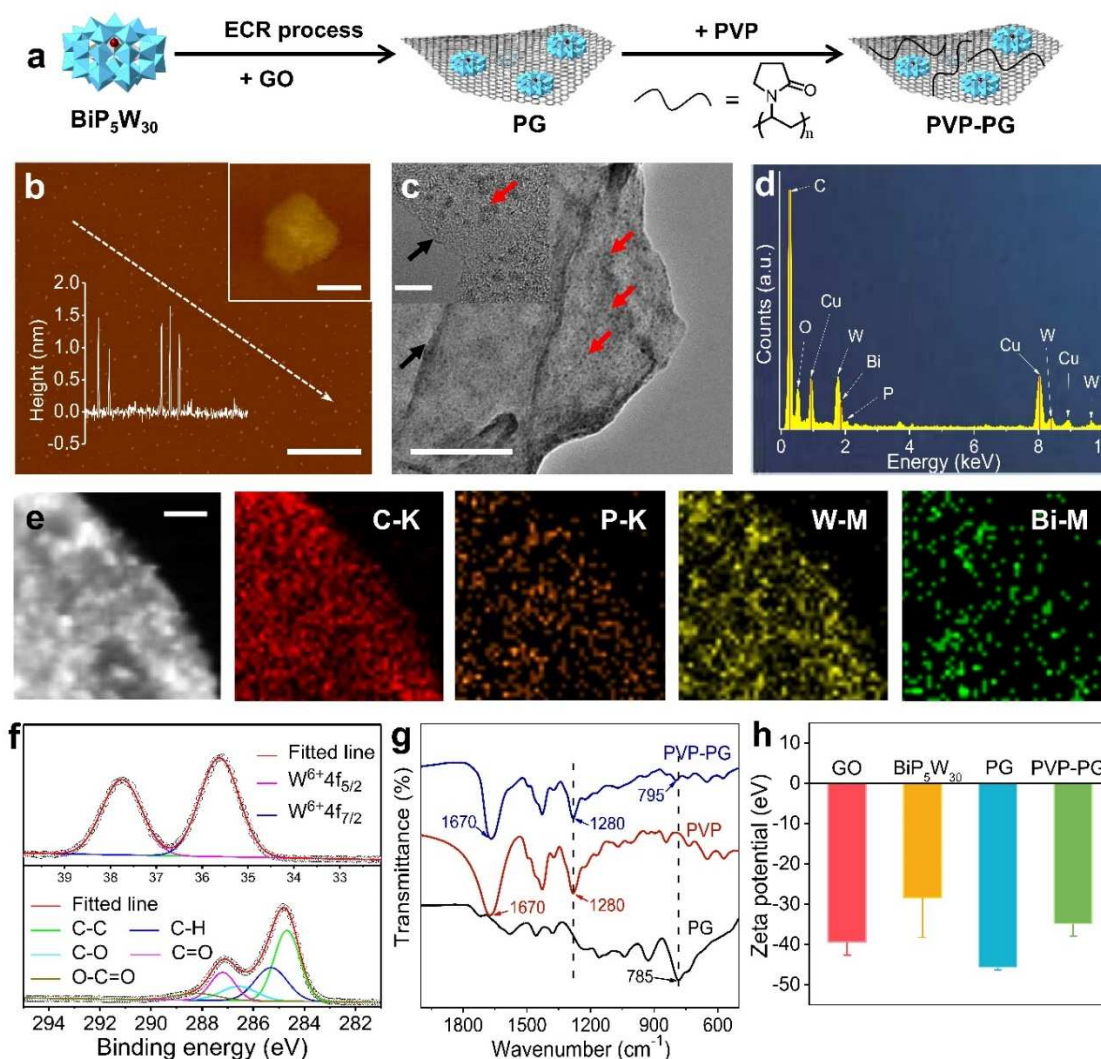


Figure 2. Synthesis and characterization of PVP-PG. (a) Schematic diagram of preparation of PVP-PG. (b) AFM image of PG. (Scale bar: 1 μm ; insert: 20 nm). (c) A TEM image of PVP-PG. Inset: an enlarged TEM image. Red arrows indicate $\text{BiP}_5\text{W}_{30}$ clusters and black arrow indicates rGO. (Scale bar: 10 nm; insert: 5 nm). (d) EDS analysis of PVP-PG. (e) The magnified TEM image and corresponding EDS element mapping images. (Scale bars: 5 nm). (f) Deconvolution of high-resolution W4f (up) and C1s (down) XPS spectra. (g) FT-IR spectra of PG, PVP, and PVP-PG, respectively. (h) Zeta potential of aqueous dispersion of GO, $\text{BiP}_5\text{W}_{30}$, PG and PVP-PG.

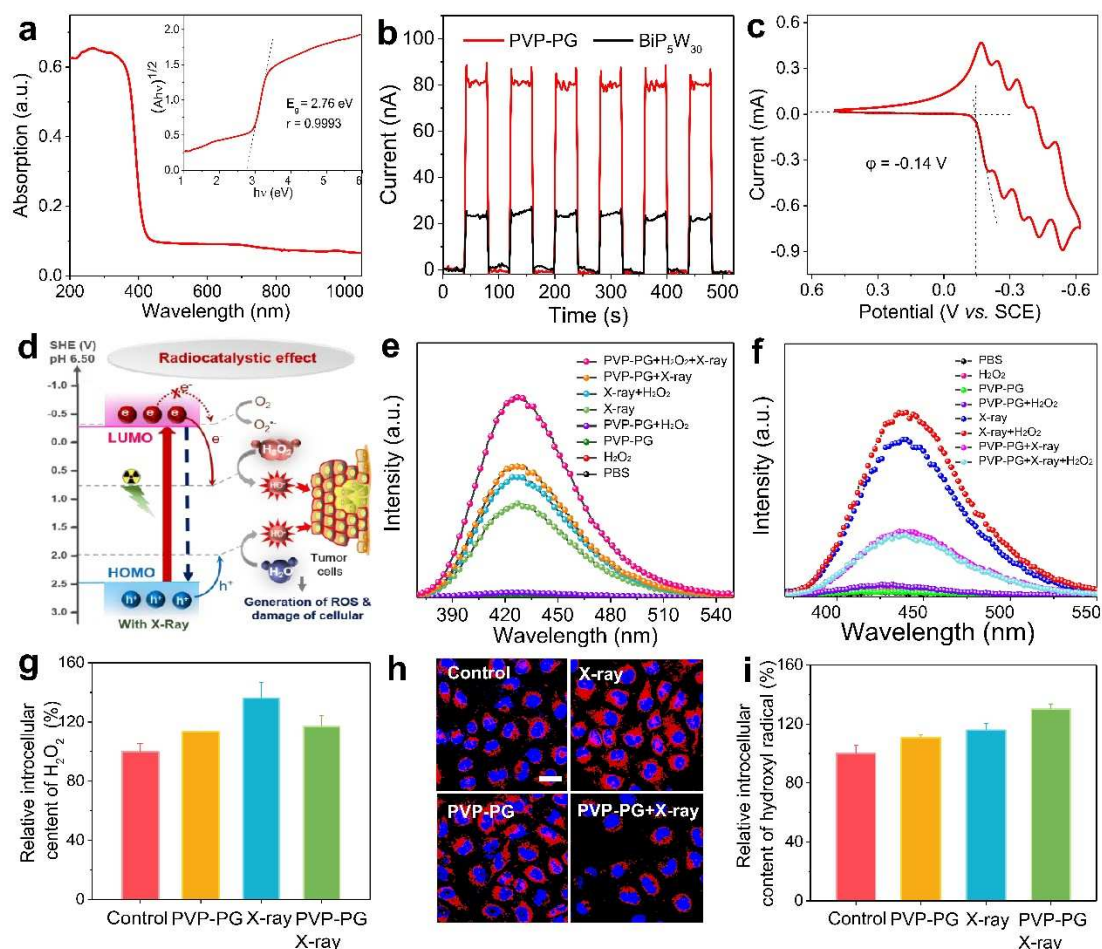


Figure 3. Radiocatalysis effect of PVP-PG. (a) Diffuse reflectance spectrum of PVP-PG. Inset: Plot transformed according to the Kubelka-Munk function versus energy of light. The dashed line is the tangent of the curve. (b) X-ray induced photocurrent with $\text{BiP}_5\text{W}_{30}$ clusters and PVP-PG, respectively, responded to the on/off of X-ray irradiation. (c) The cyclic voltammogram of PVP-PG in 0.5 M H_2SO_4 . Scan rate: 10 mV/s. (d) Schematic illustration of mechanism of PVP-PG. (e) Fluorescence spectra of TAOH after various treatments. (f) Fluorescence spectra of TAOH after various treatments in the presence of $\text{S}_2\text{O}_8^{2-}$. (g) Relative content of H_2O_2 in HeLa cells after different treatments. (h) Confocal fluorescence images of H_2O_2 in HeLa cells after different treatments. (scale bar: 50 μm). (i) Relative content of HO^\bullet in HeLa cells after different treatments.

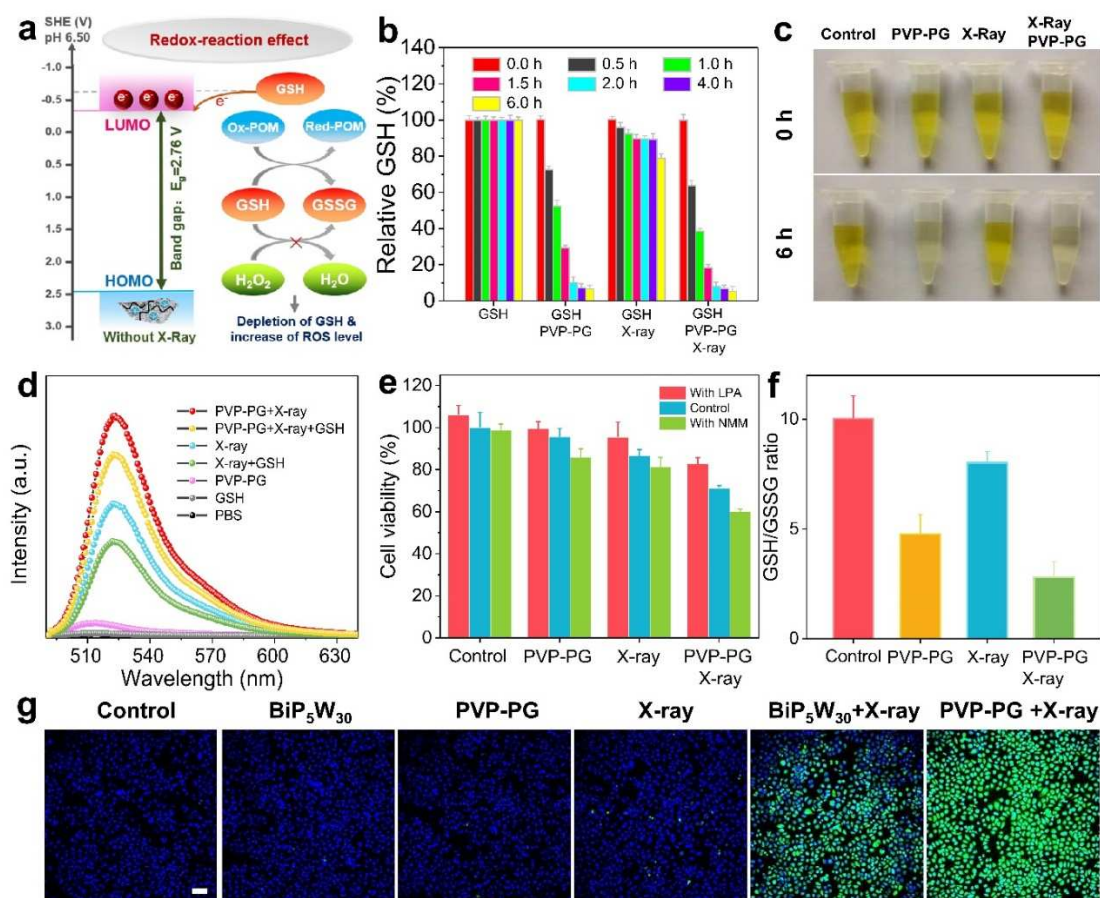


Figure 4. Redox effect of PVP-PG. (a) Schematic illustration of mechanism of PVP-PG. (b) Relative GSH amount after different treatments. (c) Photographs of color change of GSH after various treatments at 0 h and 6 h intervals under 37 °C. (d) Fluorescence spectra of DCF mixed with PVP-PG after various treatments with or without GSH. (e) *In vitro* cytotoxicity of PVP-PG with or without LPA and NMM after 24 h of different treatments. (f) Intracellular GSH/GSSG ratios in HeLa cells after different treatments. (g) Fluorescence spectra of DCF on cells after different treatments. (scale bar: 100 μ m).

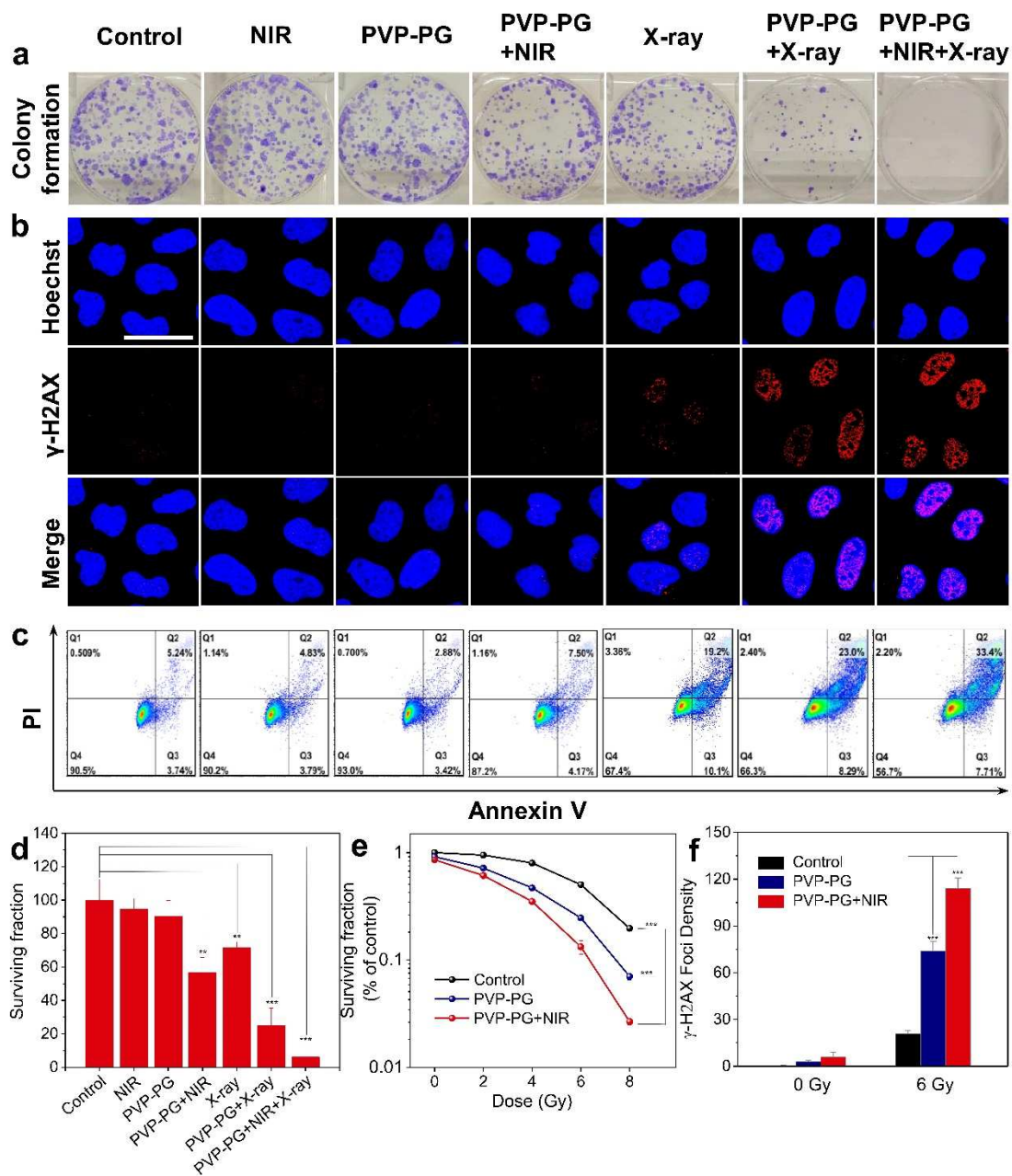


Figure 5. PTT and RT synergetic therapy *in vitro*. (a) Clonogenic assay after different treatments with clones produced by HeLa cells. (b) Qualitative representation of γ -H2AX foci formation with or without X-ray irradiation, 30 min post-irradiation (scale bar: 50 μ m). (c) Qualitative flow cytometry data plot indicating the increase in apoptosis of HeLa cells after different treatments for 24 h. (d) Responding survival fraction of HeLa cells with different treatments after 10-day colony. (e) Clonogenic survival assay of HeLa cells treated with or without PVP-PG under series of radiation doses at 0, 2, 4, 6, and 8 Gy. (f) Quantitative analysis of number of γ -H2AX foci per cell. The error bars represent the standard error of mean of at least three replicates. P values: *P < 0.05, **P < 0.01, ***P < 0.001.

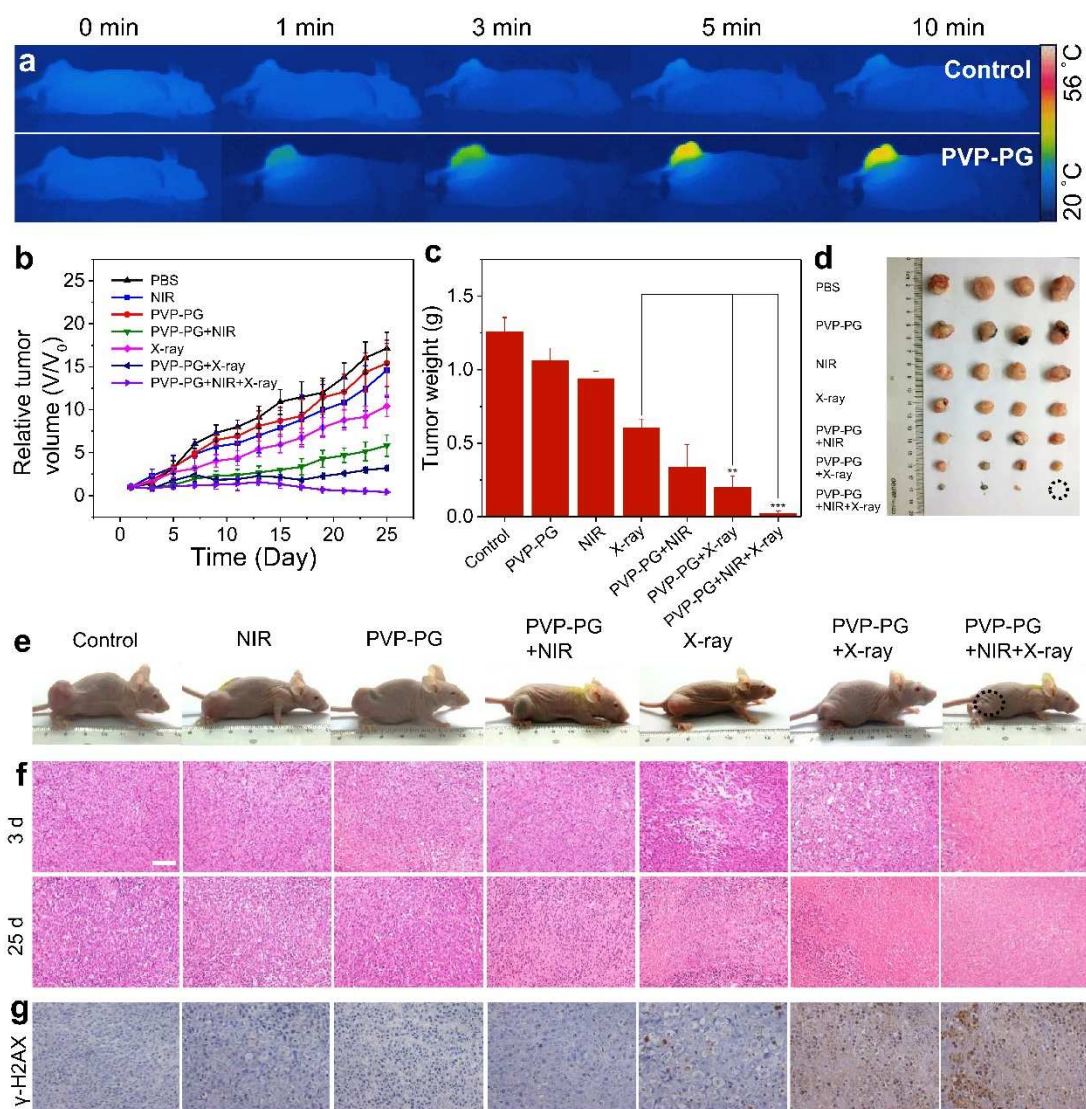
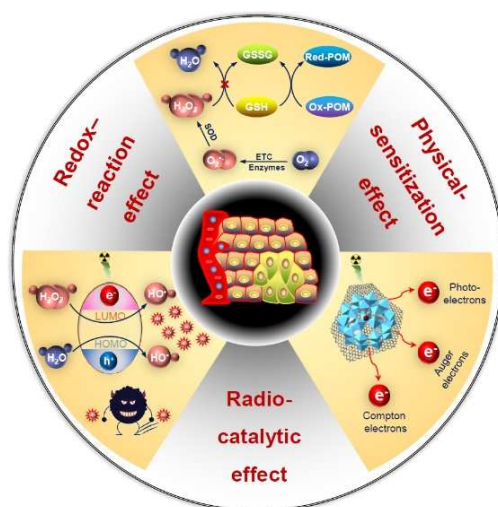


Figure 6: PTT and RT synergetic therapy *in vivo*. (a) Representative thermal response in HeLa tumor-bearing mice with intratumoral injection of PBS and PVP-PG. (b) Relative tumor volumes in HeLa tumor-bearing mice after different treatments. (n=4) (c) The tumor weights of dissected tumors from each group after 25 days of therapy. (d) Digital photographs of the dissected tumors each group. Black circle indicates the eliminated tumors. (e) Digital photographs of representative mice from each group captured on 25th day for showing the effects of different therapeutic outcomes. (f) H&E of tumor sections from each group on 3rd (up) and 25th (down) day under different treatments. (scale bar: 50 μm). (g) γ-H2AX staining representing the DNA double-strand breaks in the tumor sections on 3rd day. Damaged cells are brown and viable cells are blue. (scale bar: 50 μm), ***P < 0.001.

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The tumor microenvironment-manipulated bismuth heteropolytungstate nanoclusters could serve as a simple yet powerful radiocatalytic sensitizer for achieving tumor-specific radiotherapy by simultaneously increasing X-ray dose deposition and reversing the radio-resistance through the improvement of hypoxia environment, depletion of glutathione, and radiocatalytic conversion of overproduced hydrogen peroxide into highly toxic hydroxyl radical.